

Perovskite-based surfaces as photoanodes for an enhanced solar-driven CO₂ reduction to formate

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The photoelectrochemical reduction of CO₂ holds promise for reducing CO₂ emissions by generating value-added products with a lower energy penalty. In this process, solar energy can provide extra electrons for CO₂ reduction at the cathode in a photoanode-based reactor configuration[1]. Although TiO₂ serves as a benchmark semiconductor, its limited absorption of visible light, primarily absorbing UV light due to its wide bandgap (3.2 eV), prompts exploration of other n-type semiconductors (e.g., ZnO, BiVO₄, WO₃) for a more efficient solar irradiation harvest in the water oxidation reaction [2]. Light absorption can be further enhanced by combining these semiconductors with materials like perovskites in multi-layered surfaces, facilitating charge separation and electron mobility [3]. Thus, this work aims to develop efficient photoanodes for photo-assisted CO₂ electroreduction systems capable of generating substantial photo-generated current densities at the cathode.

To achieve this, photoanodes based on a commercial calcium titanate perovskite (CaTiO₃) and BiVO₄ layers coated onto a transparent FTO substrate by automated spray pyrolysis are proposed. Different photoanode configurations are tested, with special emphasis on the materials' position (top/bottom) and illumination conditions (back/front). The most favorable results are achieved when BiVO₄ is positioned as the top layer with back illumination [4]. In this configuration, light reaches the transparent FTO substrate, interacting first with the CaTiO₃ layer, while the BiVO₄ is in contact with the liquid electrolyte to conduct the Oxygen Evolution Reaction. Optimizing the catalytic loading of each layer reveals an optimal loading of 1 mg cm⁻² of CaTiO₃ and 3 mg cm⁻² of BiVO₄, yielding an impressive current density of -71 mA cm⁻² at -1.8 V vs. Ag/AgCl. This optimal photoanode is integrated into an electrolyzer for continuous visible light-driven CO₂ reduction in the gas phase to formate, obtaining a concentration of 63.8 g L⁻¹, with a Faradaic Efficiency of 79.1 %. These results represent a significant advancement in the development of efficient photoanode and offer advance towards future scalability of photoelectrochemical CO₂ reduction processes.

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