

Enhancing Electrochemical CO₂ Reduction: Improved Performance of a Three-Compartment Reactor for Formic Acid Production

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The World Meteorological Organization (WMO) recently announced that atmospheric carbon dioxide (CO₂) concentration has reached 420 ppm, marking a 150% surge from pre-industrial levels. In response, global initiatives like the IPCC's aim for net zero emissions by 2050 and the Paris Climate Conference (COP21) limiting the global temperature increase to below 2.0 oC, with efforts aimed at achieving 1.5 oC.

Among strategies to mitigate CO₂ emissions, electrochemical conversion presents promising avenues, particularly in transforming CO₂ into valuable products like formic acid (HCOOH) or formate (HCOO⁻). This study focuses on evaluating the performance of a novel three-compartment reactor for electrochemical CO₂ reduction, building upon prior research primarily centered on two-compartment setups. [1]

The reactor comprises a 5 cm² carbon-based (50% teflonated) cathode coated with commercial bismuth oxide (Bi₂O₃) ink, and an anode of similar dimensions featuring iridium oxide (IrO₂) catalyst on carbon paper (5% teflonated). CO₂ is introduced to the reactor via a humidified flow through a Vapor Delivery Module (VDM), while appropriate electrolytes are fed into anodic and central compartments, consisting of deionized water (Mili-Q water). [2] Various operational conditions were tested, including different CO₂ flow rates (20, 40 and 60 ml·min⁻¹), CO₂ water content (0.5, 1.5 and 3 g·h⁻¹), and current densities (45, 90 and 200 mA·cm⁻²). Results show that the three-compartment reactor achieves formic acid concentrations exceeding 140 g·L⁻¹ at 200 mA·cm⁻², with a Faradaic Efficiency of 63% for HCOOH and remarkably low energy consumption (320 kWh·kmol⁻¹) attributed to low cell potentials. [3] Another advantage of the three-compartment reactor is the production of HCOOH instead of HCOO⁻, rendering it more appealing for industrial applications.

This study presents promising avenues for industrial-scale applications, particularly in terms of reducing energy consumption, offering a potential solution to a significant barrier for wider adoption of this technology on a large scale.

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